Coupling of Methyl Ketones and Primary or Secondary Amines Leading to α -Ketoamides

Wei Wei,[†] Ying Shao,[‡] Huayou Hu,[§] Feng Zhang,[†] Chao Zhang,[†] Yuan Xu,[†] and Xiaobing Wan^{*,†}

[†]Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry, Chemical Engineering and Materials Science, Soochow University, Suzhou 215123, PR China

[‡]Jiangsu Province Key Laboratory of Fine Petrochemical Engineering, Changzhou University, Changzhou 213164, PR China

[§]Key Laboratory for Chemistry of Low-Dimensional Materials, School of Chemistry and Chemical Engineering, Huaiyin Normal University, Huaian 223300, PR China

Supporting Information

ABSTRACT: A metal-free oxidative coupling of methyl ketones and primary or secondary amines to α -ketoamides has been developed. Four intermediates, α -iodoketone, α -aminoketone, iminium intermediate, and α -hydroxy amine have been identified through a series of control experiments. The atom-economic methodology can be scaled-up, tolerates a variety of functional groups, and is operationally simple.



The iodoform reaction, discovered in 1822, is one of the oldest reactions in organic chemistry.¹ While the iodoform reaction has been widely used to prepare carboxylic acids and identify CH₃CO groups, it is surprising that few expansions have been developed since its discovery. Herein, we combine the old iodoform reaction with a modern C–H oxidation² to provide a new strategy for α -ketoamides synthesis. α -Ketoamides are prevalent structural motifs found in numerous natural products, biologically relevant molecules, and marketed drugs.³ They also serve as useful synthetic intermediates for further transformations.⁴ As a result, the synthesis of α -ketoamides has attracted considerable interest.^{5–13} Although much progress has been made in this field, the search for a simple and practical catalytic system remains a challenge for the synthetic chemist.

As shown in our retrosynthetic analysis (Scheme 1a), α ketoamides could be synthesized from α -aminoketones through

Scheme 1. Design for α -Ketoamides Synthesis

a) Retrosynthetic analysis of a-ketoamides





C-H oxidation, and α -aminoketone is the nucleophilic substitution product of amine and α -iodoketone reactants.¹² In turn, α -iodoketones can be formed by iodination of a methyl ketone.¹⁵ It is well-known that nucleophilic attack by hydroxide on a carbonyl carbon atom was involved in the catalytic cycle of the iodoform reaction (Scheme 1b). In sharp contrast, nucleophilic attack by an amine on the iodomethyl, not the carbonyl group, is a key step in this pathway. Transition metalcatalyzed C-H oxidation is of both fundamental and industrial interest, enabling clean access to many functionalized chemicals.² However, despite their remarkable potential, the toxicity and expense of transition metals is a concern. Continuing our recent studies in TBAI (tetrabutylammonium iodide)-catalyzed chemical transformation,¹⁶ we conceived of converting methyl ketone and amine to the corresponding α ketoamides using a similar metal-free approach.

RESULTS AND DISCUSSION

Initially, we investigated the room temperature reaction of acetophenone **1a** with morpholine **2a** in the presence of 50 mol % TBAI in ⁱPrOH for 13 h. The desired α -ketoamide **3a** was obtained in moderate yield (Table 1, entry 1). The use of KI under the same conditions lowered the yield (Table 1, entry 2). Finally, iodine was identified as the best catalyst for this reaction, affording α -ketoamide **3a** in excellent yield (Table 1, entry 3). No significant amount of product was observed in the absence of iodine or TBHP (Table 1, entries 4 and 5). The choice of oxidant was also crucial for this transformation. Other common oxidants (such as H₂O₂, *m*-CPBA, Oxone and O₂) suppressed the formation of desired α -ketoamide **3a** (Table 1, entries 6–9). Notably, no benzoic acid or 4-benzoylmorpholine

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Table 1. Optimization of Reaction Conditions^a

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entry	catalyst	oxidant	yield
1	TBAI	TBHP	54%
2	KI	TBHP	40%
3	I_2	TBHP	90%
4	-	TBHP	N.O. ^b
5	I_2	-	<5%
6	I_2	H_2O_2	11%
7	I_2	m-CPBA	N.O. ^b
8	I_2	Oxone	13%
9	I_2	O ₂	<5%
10	I_2	TBHP	82% ^c

^{*a*}Reaction conditions: 0.5 mmol acetophenone 1a, 2.0 mmol morpholine 2a, 50 mol % catalyst, 3.0 mmol TBHP in 2.0 mL of ^{*i*}PrOH at room temperature for 13 h. ^{*b*}Not observed. ^{*c*}100 mmol acetophenone 1a and 400 mmol morpholine 2a were used.

(nucleophilic attack by morpholine on carbonyl group) were detected as a byproduct. The high selectivity could be attributed to the strong nucleophilicity of amines and superior leaving ability of iodide. Importantly, the reaction was scaled up to 100 mmol and retained its high yield (Table 1, entry 10). The mild conditions and cheap commercial available starting materials (catalyst, oxidant and reactant) make the methodology applicable in synthetic chemistry and drug discovery.

A proposed reaction mechanism, which differs from Ji's report,^{7d} is shown in Scheme 2. In the first step, iodination of





methyl ketone **A** forms α -iodoketone **B** (step a).¹⁵ Subsequent nucleophilic substitution of amine to α -iodoketone **B** generates α -aminoketone **C** (step b).¹⁴ Further iodination produces α iodo- α -aminoketone **D** (step c). Ionization of **D** generates iminium-type intermediate **E** under the standard conditions (step d).¹⁷ Nucleophilic attack by H₂O on the resulting **E** gives the α -hydroxy amine intermediate **F** (step e).¹⁸ Finally, oxidation of **F** affords the desired α -ketoamide **G** (step f).¹⁹

A two-step one-pot synthesis of α -ketoamide **3b** was carried out to elucidate the mechanism. Treatment of 4'-bromoacetophenone **1b** with morpholine **2a** in the presence of 50 mol % iodine generated the α -aminoketone **4** (detected by LC–MS) after 1 h at ambient temperature. Subsequent addition of TBHP resulted in the desired α -ketoamide **3b** in 78% yield (Scheme 3a). When α -aminoketone **4** was subjected to the standard conditions, α -ketoamide **3b** was obtained in 81% yield (Scheme 3b). Notably, only trace amount of α -ketoamide 3b was observed in the absence of iodine (Scheme 3c). We suspected α -aminoketone 4 is generated in situ by the nucleophilic substitution of morpholine 2a on 1-(4-bromophenyl)-2-iodoethanone 5. However, no 1-(4-bromophenyl)-2-iodoethanone 5 was observed by LC-MS during the reaction, which might be owing to the fast reaction of morpholine 2a and 1-(4-bromophenyl)-2-iodoethanone 5. As anticipated, when 1-(4-bromophenyl)-2-iodoethanone 5 was used in the standard conditions, a 76% yield of α -ketoamide 3b was obtained (Scheme 3c). On the basis of the above results, both α -iodoketones and α -aminoketones are reaction intermediates in the formation of α -ketoamides.

Further investigations into the mechanism were performed. As shown in Scheme 4a, hydroxide anion is generated in situ from TBHP in the standard conditions.²⁰ Consequently, when an ¹⁸O-labeled reaction was carried out, both **3b**-¹⁸O and **3b** were obtained (Scheme 4b). Therefore, we believed the nucleophilic attack of a water molecule on the iminium-type intermediate, not by TBHP, is involved in the catalytic cycle. Notably, benzoylformaldehyde was also a suitable reaction partner, which suggested the α -hydroxy amine served as an intermediate in the transformation (Scheme 4c).

Next, a variety of substituted aryl methyl ketones were subjected to the standard conditions. Representative results are summarized in Figure 1. A host of aryl methyl ketones with ortho, meta, or para electron-donating or electron-withdrawing substituents reacted smoothly with morpholine 2a in moderate to excellent yields. Interestingly, the presence of bromo and chloro substituents on the aromatic groups did not interfere with the C-H oxidation process, affording products 3b, 3e and 3m that could be further functionalized by transition metalcatalyzed cross-coupling reactions. The reaction was also compatible with fluoro (3d), benzylic C-H (3f), trifluoromethyl (3g), nitriles (3h), esters (3i), alkyne (3j), sulfonate (3k), nitro (3l and 3o), and ether (3p) groups. Even oxidative sensitive groups, such as C,C double bond (3i) and an unprotected hydroxyl group (3n), were tolerated in the transformation. Finally, steric effects did not significantly influence the reactivity of the reaction, with product 3q being obtained in high yield.

Most recently, Ji and co-workers have developed an elegant copper-catalyzed oxidative coupling of methyl ketones and amines using molecular oxygen.^{7d} In their work, the ketones are limited primarily to aryl methyl ketones. To further showcase the potential of our methodology, ketones with heteroaryl substituents were submitted to the optimized conditions. As shown in Figure 2, both 2-acetylthiophene and 2-acetyl-5-chlorothiophene were compatible reaction partners for this transformation, leading to products **6a** and **6b** in high yields. Thiazole (**6c**), pyrazine (**6d**) and furan (**6h**) were also tolerated in the α -ketoamide formation reaction. Finally, 2-acetylpyridine, 3-acetylpyridine, and 4-acetylpyridine afforded the desired α -ketoamide **6e**-**6g** in moderate yields.

To further define the scope of the α -ketoamide synthesis, we next applied this process to a series of amines as shown in Figure 3. Pyrrolidine and piperidine reacted well with acetophenone 1a to afford 7a and 7c in 73 and 64% yields, respectively. Acyclic amines also worked well and gave the corresponding α -ketoamides in moderate to excellent yields (7b, 7d and 7h). A variety of functional groups, including ether (7h), pyridine (7i), ester (71), and tertiary amine (7i), were unaffected in the reaction, and corresponding α -ketoamides

Scheme 3. Investigations into the Reaction Mechanism



Scheme 4. Investigation into the Reaction Mechanism



were obtained in high yields. Synthetically useful benzyl and tetrahydroisoquinoline are tolerated in the transformation, giving 7e-7g in good yields. Notably, free hydroxyl (7k) and removable BocNH group (7j) were inert in the reaction, leading to expected α -ketoamides in satisfactory yields.

Next, primary amines were used as reaction partners for the synthesis of α -ketoamides. As shown in Figure 4, these reactions also tolerate a variety of aromatic substituents (electron-rich, electron-poor, and heteroaromatic), while potentially sensitive functional groups (double bond and benzylic C–H bond) are also unaffected by the reaction conditions.

CONCLUSIONS

In summary, we have described an oxidative coupling of methyl ketones with primary and secondary amines to give α -ketoamides in moderate to excellent yields. The reaction makes direct use of simple and abundant starting materials without requiring transition metal catalysts. In view of the wide functional group tolerance, the ease of conducting such reactions, and the mild reaction conditions, we envision this protocol will be widely adapted in synthetic chemistry.

EXPERIMENTAL SECTION

General Methods. All manipulations were carried out under air atmosphere. Column chromatography was generally performed on silica gel (300–400 mesh), and reactions were monitored by thin layer chromatography (TLC) using UV light to visualize the course of the reactions. The ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) data were recorded on 400 M spectrometers using CDCl₃ as solvent at room temperature. The chemical shifts (δ) are reported in ppm and coupling constants (J) in Hz. ¹H NMR spectra were recorded with tetramethylsilane ($\delta = 0.00$ ppm) as internal reference; ¹³C NMR spectra were recorded with CDCl₃ ($\delta = 77.00$ ppm) as internal reference.

General procedures for amides **3a–3q**, **6a–6h**, **7a–7l**, and **8a–8h**: Methyl ketone (0.5 mmol), amine (2.0 mmol), I₂ (0.25 mmol, 50 mol %), TBHP (6.0 equiv, 70% aqueous solution 401 uL), and 2.0 mL of ⁱPrOH were added to a tube under air. The reaction mixture was stirred at room temperature or 0 °C for 13 h. The reaction mixture was quenched with saturated Na₂S₂O₃ solution, extracted repeatedly with ethyl acetate, and dried over Na₂SO₄. Removal of the organic solvent in a vacuum followed by flash silica gel column chromatographic purification (petroleum/ethyl acetate) afforded product.

1-Morpholino-2-phenylethane-1,2-dione (3a). Purified by flash column chromatography (EtOAc:hexane 1:5). Yellow oil (99 mg, 90% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.98–7.95 (m, 2H),

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Figure 1. Scope of methyl ketones. Reaction conditions: 0.5 mmol ketones 1, 2.0 mmol morpholine 2a, 50 mol % iodine, 3.0 mmol TBHP in 2.0 mL of ⁱPrOH at room temperature for 13 h.



Figure 2. Scope of heteroaryl ketones. Reaction conditions: 0.5 mmol ketones 1, 2.0 mmol morpholine 2a, 50 mol % iodine, 3.0 mmol TBHP in 2.0 mL of ⁱPrOH at room temperature for 13 h.

7.68–7.64 (m, 1H), 7.55–7.51 (m, 2H), 3.81 (t, J = 4.0 Hz, 2H), 3.79 (t, J = 4.0 Hz, 2H), 3.66 (t, J = 4.0 Hz, 2H), 3.39 (t, J = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.0, 165.2, 134.8, 132.8, 129.4, 128.9, 66.5, 66.4, 46.0, 41.4; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₂H₁₄NO₃ 220.0968, found 220.0975; IR (KBr, cm⁻¹) ν 1681, 1645.

Scale-up of Product 3a. Acetophenone (100.0 mmol), morpholine (400.0 mmol), I_2 (50.0 mmol, 50 mol %), TBHP (6.0 equiv, 70% aqueous solution 80.2 mL), and 400.0 mL of ⁱPrOH, were added to a flask under air. The reaction mixture was stirred at room temperature for 13 h. Then, the reaction mixture was quenched with saturated $Na_2S_2O_3$ solution and then extracted three times with ethyl acetate. Removal of organic solvent got the raw product. Finally, flash silica gel column chromatographic purification (petroleum/ethyl acetate) afforded product 3a in 82% yield.

1-(4-Bromophenyl)-2-morpholinoethane-1,2-dione (3b). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (122 mg, 82% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.83 (d, *J* = 8.6 Hz, 2H), 7.68 (d, *J* = 8.6 Hz, 2H), 3.79 (t, *J* = 4.0 Hz, 4H), 3.67 (t, *J* = 4.0 Hz, 2H), 3.38 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 189.8, 164.7, 132.3, 131.7, 130.9, 130.3, 66.5, 66.4, 46.1, 41.5; HRMS (TOF) m/z [M]⁺ Calcd for C₁₂H₁₂⁷⁹BrNO₃ 297.0001, found 297.0007; Calcd for C₁₂H₁₂⁸¹BrNO₃ 298.9980, found 299.0022; IR (KBr, cm⁻¹) ν 1676, 1628.

1-Morpholino-2-(naphthalen-1-yl)ethane-1,2-dione (3c). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (112 mg, 83% yield): ¹H NMR (CDCl₃, 400 MHz) δ 9.25 (d, J = 8.7 Hz, 1H), 8.14 (d, J = 8.2 Hz, 1H), 8.05–8.03 (m, 1H), 7.93 (d, J = 8.1 Hz, 1H), 7.73–7.69 (m, 1H), 7.63–7.55 (m, 2H), 3.88–3.78 (m, 4H), 3.67 (t, J = 4.0 Hz, 2H), 3.44 (t, J = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,

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Figure 3. Scope of amines. Reaction conditions: 0.5 mmol acetophenone 1a, 2.0 mmol amines 2, 50 mol % iodine, 3.0 mmol TBHP in 2.0 mL of ⁱPrOH at room temperature for 13 h.



Figure 4. Scope of primary amines. Reaction conditions: 0.5 mmol methyl ketone 1, 2.0 mmol amines 2, 50 mol % iodine, 3.0 mmol TBHP in 2.0 mL of ⁱPrOH at 0 °C for 13 h.

75 MHz) δ 193.4, 165.8, 136.0, 134.3, 133.8, 130.6, 129.3, 128.6, 128.1, 126.9, 125.5, 124.3, 66.41, 66.40, 46.1, 41.5; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₆H₁₆NO₃ 270.1125, found 270.1136; IR (KBr, cm⁻¹) ν 1669, 1630.

1-(4-Fluorophenyl)-2-morpholinoethane-1,2-dione (3d). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (97 mg, 82% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.03–7.99 (m, 2H), 7.22–7.18 (m, 2H), 3.80–3.77 (m, 4H), 3.67 (t, *J* = 4.0 Hz, 2H), 3.39 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 189.2, 168.3, 164.9, 132.4, 132.3, 129.4, 116.4, 116.1, 66.5, 66.4, 46.1, 41.5; HRMS (TOF) *m*/*z* [M]⁺ Calcd for C₁₂H₁₂FNO₃ 237.0801, found 237.0809; IR (KBr, cm⁻¹) *ν* 1672, 1633.

1-(4-Chlorophenyl)-2-morpholinoethane-1,2-dione (3e). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (93 mg, 73% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.93–7.90 (m, 2H), 7.52–7.50 (m, 2H), 3.82–3.77 (m, 4H), 3.67 (t, J = 4.0 Hz, 2H), 3.39 (t, J = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 189.5, 164.7, 141.4, 131.3, 130.9, 129.3, 66.5, 66.4, 46.1, 41.5; HRMS (TOF) m/z

 $[M + H]^+$ Calcd for $C_{12}H_{13}^{35}$ ClNO₃ 254.0578, found 254.0586; IR (KBr, cm⁻¹) ν 1676,1628.

1-Morpholino-2-p-tolylethane-1,2-dione (3f). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (84 mg, 72% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.85 (d, *J* = 8.8 Hz, 2H), 7.32 (d, *J* = 8.8 Hz, 2H), 3.80–3.76 (m, 4H), 3.65 (t, *J* = 4.0 Hz, 2H), 3.37 (t, *J* = 4.0 Hz, 2H), 2.44 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 190.8, 165.5, 146.1, 130.4, 129.7, 129.6, 66.53, 66.46, 46.1, 41.4, 21.8; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₃H₁₆NO₃ 234.1125, found 234.1134; IR (KBr, cm⁻¹) ν 1675, 1645.

1-Morpholino-2-(4-(trifluoromethyl)phenyl)ethane-1,2dione (3g). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (102 mg, 71% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.10 (d, *J* = 8.0 Hz, 2H), 7.79 (d, *J* = 8.0 Hz, 2H), 3.82–3.79 (m, 4H), 3.68 (t, *J* = 4.0 Hz, 2H), 3.41 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 189.6, 164.4, 135.9, 135.6, 135.5, 129.9, 126.03, 126.0, 125.96, 125.9, 124.6, 121.8, 66.6, 66.5, 46.2, 41.7; HRMS (TOF) *m*/*z*

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 $[M + H]^+$ Calcd for $C_{13}H_{13}F_3NO_3$ 288.0842, found 288.0839; IR (KBr, cm⁻¹) ν 1681, 1630.

4-(2-Morpholino-2-oxoacetyl)benzonitrile (3h). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (65 mg, 53% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.09 (d, *J* = 8.0 Hz, 2H), 7.84 (d, *J* = 8.0 Hz, 2H), 3.83–3.79 (m, 4H), 3.69 (t, *J* = 4.0 Hz, 2H), 3.41 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 188.9, 163.9, 135.8, 132.7, 129.9, 117.7, 117.4, 66.5, 66.4, 46.1, 41.7; HRMS (TOF) *m*/*z* [M + Na]⁺ Calcd for C₁₃H₁₂N₂NaO₃ 267.0740, found 267.0738; IR (KBr, cm⁻¹) ν 1686, 1643.

(É)-Methyl 3-(4-(2-morpholino-2-oxoacetyl)phenyl)acrylate (3i). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (93 mg, 61% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.99 (d, *J* = 8.2 Hz, 2H), 7.72 (d, *J* = 16.1 Hz, 1H), 7.66 (d, *J* = 8.2 Hz, 2H), 6.56 (d, *J* = 16.1 Hz, 1H), 3.84 (s, 3H), 3.81–3.80 (m, 4H), 3.67 (t, *J* = 4.0 Hz, 2H), 3.40 (t, *J* = 4.0 Hz, 2H),; ¹³C NMR (CDCl₃, 100 MHz) δ 189.9, 166.4, 164.8, 142.5, 140.2, 133.7, 129.9, 128.3, 121.1, 66.5, 66.4, 51.7, 46.0, 41.4; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₆H₁₈NO₅ 304.1179, found 304.1179; IR (KBr, cm⁻¹) ν 1725, 1667, 1641.

1-Morpholino-2-(4-(phenylethynyl)phenyl)ethane-1,2-dione (3j). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (121 mg, 76% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.95 (d, *J* = 8.3 Hz, 2H), 7.66 (d, *J* = 8.3 Hz, 2H), 7.57–7.55 (m, 2H), 7.39–7.38 (m, 3H), 3.81–3.80 (m, 4H), 3.67 (t, *J* = 4.0 Hz, 2H), 3.40 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 190.1, 165.0, 132.0, 131.9, 131.7, 130.0, 129.5, 129.0, 128.4, 122.2, 94.0, 88.3, 66.6, 66.5, 46.1, 41.5; HRMS (TOF) *m*/*z* [M]⁺ Calcd for C₂₀H₁₇NO₃ 319.1208, found 319.1223; IR (KBr, cm⁻¹) ν 1673, 1642.

4-(2-Morpholino-2-oxoacetyl)phenyl 4-methylbenzenesulfonate (**3k**). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (148 mg, 76% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.94–7.91 (m, 2H), 7.75–7.71 (m, 2H), 7.35–7.33 (m, 2H), 7.18– 7.14 (m, 2H), 3.81–3.76 (m, 4H), 3.65 (t, *J* = 4.0 Hz, 2H), 3.38 (t, *J* = 4.0 Hz, 2H), 2.46 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 189.2, 164.6, 154.0, 145.9, 131.7, 131.4, 129.9, 128.2, 122.8, 66.5, 66.4, 46.1, 41.5, 21.6; HRMS (TOF) *m*/*z* [M]⁺ Calcd for C₁₉H₁₉NO₆S 389.0933, found 389.0930; IR (KBr, cm⁻¹) ν 1682, 1646.

1-Morpholino-2-(4-nitrophenyl)ethane-1,2-dione (3l). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (73 mg, 55% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.37 (d, *J* = 8.0 Hz, 2H), 8.17 (d, *J* = 8.0 Hz, 2H), 3.83–3.81 (m, 4H), 3.70 (t, *J* = 4.0 Hz, 2H), 3.43 (t, *J* = 4.0 Hz, 2H), ¹³C NMR (CDCl₃, 75 MHz) δ 188.6, 163.9, 151.0, 137.3, 130.7, 124.0, 66.6, 66.4, 46.2, 41.8; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₂H₁₃N₂O₅ 265.0819, found 265.0824; IR (KBr, cm⁻¹) ν 1686, 1636.

1-(3-Bromophenyl)-2-morpholinoethane-1,2-dione (3m). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (112 mg, 75% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.98–7.95 (m, 2H), 7.67–7.64 (m, 1H), 7.55–7.51 (m, 2H), 3.82–3.77 (m, 4H), 3.66 (t, *J* = 4.0 Hz, 2H), 3.39 (t, *J* = 4.0 Hz, 2H),; ¹³C NMR (CDCl₃, 100 MHz) δ 189.3, 164.4, 137.5, 134.6, 132.1, 130.5, 128.2, 123.1, 66.5, 66.4, 46.1, 41.5; HRMS (TOF) *m*/*z* [M]⁺ Calcd for C₁₂H₁₂⁷⁹BrNO₃ 297.0001, found 297.0013; C₁₂H₁₂⁸¹BrNO₃ 298.9980, found 298.9995; IR (KBr, cm⁻¹) *ν* 1677, 1634.

1-(3-Hydroxyphenyl)-2-morpholinoethane-1,2-dione (3n). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (60 mg, 51% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.94 (brs, 1H), 7.43–7.41 (m, 1H), 7.36–7.29 (m, 2H), 7.09–7.06 (m, 1H), 3.81–3.77 (m, 4H), 3.66 (t, *J* = 4.0 Hz, 2H), 3.37 (t, *J* = 4.0 Hz, 2H), ¹³C NMR (CDCl₃,100 MHz) δ 190.9, 165.9, 157.1, 133.7, 130.3, 122.9, 121.5, 115.6, 66.5, 66.4, 46.3, 41.7; HRMS (TOF) *m*/*z* [M + Na]⁺ Calcd for C₁₂H₁₃NNaO₄ 258.0737, found 258.0735; IR (KBr, cm⁻¹) ν 1681, 1630.

1-Morpholino-2-(3-nitrophenyl)ethane-1,2-dione (30). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (60 mg, 45% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.79–8.78 (m, 1H), 8.51–8.31 (m, 2H), 7.78–7.74 (m, 1H), 3.84–3.82 (m, 4H), 3.71 (t, *J* = 4.0 Hz, 2H), 3.46 (t, *J* = 4.0 Hz, 2H),; ¹³C NMR (CDCl₃,100 MHz) δ 188.1, 163.8, 148.5, 135.1, 134.4, 130.3, 128.7, 124.4, 66.6, 66.5, 46.2,

41.9; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₂H₁₃N₂O₅ 265.0819, found 265.0818; IR (KBr, cm⁻¹) ν 1685, 1645.

1-(3-Methoxyphenyl)-2-morpholinoethane-1,2-dione (3p). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (107 mg, 86% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.52–7.50 (m, 2H), 7.45–7.41 (m, 1H), 7.22–7.19 (m, 1H), 3.87 (s, 3H), 3.81–3.78 (m, 4H), 3.66 (t, *J* = 4.0 Hz, 2H), 3.38 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 190.9, 165.2, 159.9, 134.1, 130.0, 122.6, 121.6, 112.6, 66.5 66.4, 55.4, 46.1, 41.4. HRMS (TOF) *m/z* [M + H]⁺ Calcd for C₁₃H₁₆NO₄ 250.1074, found 250.1074; IR (KBr, cm⁻¹) ν 1681, 1646.

1-Morpholino-2-o-tolylethane-1,2-dione (3q). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (100 mg, 86% yield): ¹H NMR (CDCl₃, 400 MHz) *δ* 7.73–7.71 (m, 1H), 7.52–7.48 (m, 1H), 7.35–7.31 (m, 2H), 3.80–3.76 (m, 4H), 3.67 (t, *J* = 4.0 Hz, 2H), 3.40 (t, *J* = 4.0 Hz, 2H), 2.67 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) *δ* 192.9, 166.0, 141.4, 133.7, 132.51, 132.48, 131.2, 126.0, 66.42, 66.39, 46.0, 41.4, 21.6; HRMS (TOF) *m/z* [M]⁺ Calcd for C₁₃H₁₅NO₃ 233.1052, found 233.1054; IR (KBr, cm⁻¹) *ν* 1673, 1629.

1-Morpholino-2-(thiophen-2-yl)ethane-1,2-dione (6a). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (105 mg, 93% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.86–7.82 (m, 2H), 7.21–7.19 (m, 1H), 3.81–3.75 (m, 4H), 3.69 (t, *J* = 4.0 Hz, 2H), 3.50 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 182.7, 164.1, 140.0, 136.7, 136.1, 128.6, 66.6, 66.4, 46.2, 41.7; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₀H₁₂NO₃S 226.0532, found 226.0532; IR (KBr, cm⁻¹) ν 1651.

1-(5-Chlorothiophen-2-yl)-2-morpholinoethane-1,2-dione (6b). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (106 mg, 82% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.67 (d, *J* = 4.1 Hz, 1H), 7.04 (d, *J* = 4.1 Hz, 1H), 3.78–3.74 (m, 4H), 3.69 (t, *J* = 4.0 Hz, 2H), 3.53 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 181.2, 163.2, 142.7, 138.3, 135.9, 128.1, 66.6, 66.4, 46.3, 41.9; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₀H₁₀ClNO₃S 259.0070, found 259.0070; IR (KBr, cm⁻¹) ν 1640.

1-Morpholino-2-(thiazol-2-yl)ethane-1,2-dione (6c). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (59 mg, 52% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.15 (d, *J* = 3.0 Hz, 1H), 7.87 (d, *J* = 3.0 Hz, 1H), 3.84–3.78 (m, 4H), 3.71 (t, *J* = 4.0 Hz, 2H), 3.42 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 182.8, 163.8, 163.1, 146.0, 127.8, 66.34, 66.29, 46.1, 41.7; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₉H₁₀N₂O₃S 226.0412, found 226.0411; IR (KBr, cm⁻¹) ν 1686, 1647.

1-(3-Ethylpyrazin-2-yl)-2-morpholinoethane-1,2-dione (6d). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (64 mg, 51% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.71 (d, *J* = 2.2 Hz, 1H), 8.55 (d, *J* = 2.2 Hz, 1H), 3.84–3.77 (m, 4H), 3.74–3.72 (m, 2H), 3.41–3.40 (m, 2H), 3.27 (q, *J* = 8.0 Hz, 2H), 1.35 (t, *J* = 8.0 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.5, 166.4, 160.9, 147.0, 144.0, 141.3, 66.43, 66.38, 46.1, 41.6, 28.1, 12.6; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₂H₁₅N₃O₃ 249.1133, found 249.1133; IR (KBr, cm⁻¹) ν 1701, 1643.

1-Morpholino-2-(pyridin-2-yl)ethane-1,2-dione (6e). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (76 mg, 69% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.77–8.76 (m, 1H), 8.12–8.10 (m, 1H), 7.95–7.91 (m, 1H), 7.58–7.55 (m, 1H), 3.84–3.80 (m, 4H), 3.70 (t, J = 4.0 Hz, 2H), 3.36 (t, J = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 191.5, 166.3, 150.9, 149.8, 137.2, 128.1, 123.0, 66.3, 46.0, 41.4; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₁H₁₃N₂O₃ 221.0921, found 221.0921; IR (KBr, cm⁻¹) ν 1702, 1641.

1-Morpholino-2-(pyridin-3-yl)ethane-1,2-dione (6f). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (57 mg, 52% yield): ¹H NMR (CDCl₃, 400 MHz) δ 9.16–8.86 (m, 2H), 8.30–8.27 (m, 1H), 7.51–7.48 (m, 1H), 3.82–3.81 (m, 4H), 3.70 (t, *J* = 4.0 Hz, 2H), 3.44 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 189.3, 164.0, 154.7, 151.0, 136.6, 128.6, 123.8, 66.5, 66.4, 46.1, 41.7; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₁H₁₃N₂O₃ 221.0921, found 221.0918; IR (KBr, cm⁻¹) ν 1683, 1646.

1-Morpholino-2-(pyridin-4-yl)ethane-1,2-dione (6g). Purified by column chromatography (EtOAc:hexane 1:5). Yellow oil (48 mg, 43% yield): ¹H NMR (CDCl₃, 400 MHz) δ ¹H NMR (400 MHz, CDCl₃) δ 8.89 (d, *J* = 5.8 Hz, 2H), 7.78 (d, *J* = 5.8 Hz, 2H), 3.83–3.79 (m, 4H), 3.70 (t, *J* = 4.0 Hz, 2H), 3.42 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 189.7, 163.8, 151.2, 138.9, 122.0, 66.7, 66.6, 46.2, 41.8; HRMS (TOF) *m*/*z* [M]⁺ Calcd for C₁₁H₁₂N₂O₃ 220.0848, found 220.0848; IR (KBr, cm⁻¹) ν 1675, 1643.

1-(Benzofuran-2-yl)-2-morpholinoethane-1,2-dione (6h). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (65 mg, 71% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.76–7.74 (m, 2H), 7.63–7.61 (m, 1H), 7.56–7.52 (m, 1H), 7.37–7.34 (m, 1H), 3.81–3.78 (m, 4H), 3.71 (t, *J* = 4 Hz, 2H), 3.54 (t, *J* = 4 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ 179.7, 163.6, 156.4, 149.8, 129.6, 126.6, 124.4, 123.8, 118.8, 112.6, 66.7, 66.5, 46.3, 42.0; HRMS (TOF) *m*/*z* [M + 1]⁺ Calcd for C₁₄H₁₄NO₄ 260, found 260; IR (KBr, cm⁻¹) *ν* 1655.

1-Phenyl-2-(pyrrolidin-1-yl)ethane-1,2-dione (7a). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (74 mg, 73% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.00–7.98 (m, 2H), 7.66–7.62 (m, 1H), 7.53–7.49 (m, 2H), 3.66 (t, *J* = 6.5 Hz, 2H), 3.42 (t, *J* = 6.5 Hz, 2H), 1.98–1.92 (m, 4H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.3, 164.7, 134.4, 132.5, 129.5, 128.7, 46.4, 44.9, 25.6, 23.7; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₂H₁₄NO₂ 204, found 204; IR (KBr, cm⁻¹) ν 1675, 1634.

N,*N*-Dimethyl-2-oxo-2-phenylacetamide (7b). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (57 mg, 64% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.95–7.93 (m, 2H), 7.66–7.63 (m, 1H), 7.53–7.49 (m, 2H), 3.11 (s, 3H), 2.95 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.7, 166.9, 134.6, 132.8, 129.4, 128.9, 36.9, 33.8; HRMS (TOF) m/z [M + Na]⁺ Calcd for C₁₀H₁₁NNaO₂ 200.0682, found 200.0670; IR (KBr, cm⁻¹) ν 1681, 1651.

1-Phenyl-2-(piperidin-1-yl)ethane-1,2-dione (7c). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (70 mg, 64% yield): ¹H NMR (CDCl₃, 300 MHz) δ 7.96–7.94 (m, 2H), 7.66–7.63 (m, 1H), 7.54–7.50 (m, 2H), 3.71 (t, *J* = 5.0 Hz, 2H), 3.30 (t, *J* = 5.0 Hz, 2H), 1.72–1.68 (m, 4H), 1.57–1.53 (m, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 191.9, 165.4, 134.6, 133.2, 129.5, 128.9, 47.0, 42.1, 26.1, 25.4, 24.3; ESI (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₃H₁₆NO₂ 218, found 218; IR (KBr, cm⁻¹) ν 1671, 1645.

N,*N*-Dihexyl-2-oxo-2-phenylacetamide (7d). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (144 mg, 91% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.95–7.93 (m, 2H), 7.65–7.62 (m, 1H), 7.52–7.49 (m, 2H), 3.49 (t, *J* = 8.0 Hz, 2H), 3.15 (t, *J* = 8.0 Hz, 2H), 1.70–1.64 (m, 2H), 1.56–1.52 (m, 2H), 1.41–1.33 (m, 6H), 1.21–1.14 (m, 6H), 0.92 (t, *J* = 6.6 Hz, 3H), 0.82 (t, *J* = 6.6 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.5, 166.9, 134.4, 133.2, 129.5, 128.8, 47.5, 44.1, 31.4, 31.1, 28.4, 27.2, 26.6, 26.1, 22.5, 22.3, 13.9, 13.8; HRMS (TOF) *m*/*z* [M + Na]⁺ Calcd for C₂₀H₃₁NNaO₂ 340.2247, found 340.2228; IR (KBr, cm⁻¹) ν 1682, 1644.

N,*N*-Dibenzyl-2-oxo-2-phenylacetamide (7e). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (148 mg, 90% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.01–7.99 (m, 2H), 7.67–7.63 (m, 1H), 7.54–7.50 (m, 2H), 7.40–7.36 (m, 2H), 7.35–7.30 (m, 6H), 7.26–7.23 (m, 2H), 4.63 (s, 2H), 4.28 (s, 2H). ¹³C NMR (CDCl₃, 100 MHz) δ 191.2, 167.3, 135.8, 134.7, 134.6, 133.2, 129.6, 128.9, 128.8, 128.7, 128.6, 128.1, 128.0, 127.8, 50.0, 45.9; HRMS (TOF) m/z [M + Na]⁺ Calcd for C₂₂H₁₉NNaO₂ 352.1308, found 352.1310; IR (KBr, cm⁻¹) ν 1679, 1631.

N-Benzyl-N-methyl-2-oxo-2-phenylacetamide (7f). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (85 mg, 67% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.00–7.95 (m, 2H), 7.64–7.60 (m, 1H), 7.51–7.47 (m, 2H), 7.37–7.36 (m, 2H), 7.32–7.24 (m, 3H), 4.72 (s, 1.08 H), 4.37 (s, 0.91 H), 2.98 (s, 1.37 H), 2.82 (s, 1.63 H). ¹³C NMR (CDCl₃, 100 MHz) δ 191.4, 191.3, 167.1, 166.9, 135.5, 134.7, 134.6, 134.6, 132.9, 132.8, 129.5, 129.4, 128.9, 128.8, 128.6, 128.6, 128.1, 128.0, 127.7, 127.6, 53.2, 49.5, 34.2, 31.1; HRMS (TOF) *m*/*z* [M + Na]⁺ Calcd for C₁₆H₁₅NNaO₂ 276.0995, found 276.1012; IR (KBr, cm⁻¹) ν 1681, 1645.

tert-Butyl 4-(2-oxo-2-phenylacetyl)piperazine-1-carboxylate (**7g**). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (81 mg, 61% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.99–7.93

(m, 2H), 7.66–7.59 (m, 1H), 7.52–7.45 (m, 2H), 7.25–7.18 (m, 2.5H), 7.12–7.10 (m, 1H), 6.92–6.91 (m, 0.5H), 4.90 (s, 1.28H), 4.53 (s, 0.72H), 3.98 (t, J = 6.0 Hz, 0.7H), 3.60 (t, J = 6.0 Hz, 1.30H), 2.99 (t, J = 6.0 Hz, 0.70H), 2.84 (t, J = 6.0 Hz, 1.30H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.4, 191.2, 165.9, 165.6, 134.73, 134.68, 134.0, 133.3, 132.9, 132.8, 131.6, 131.4, 129.6, 129.5, 128.93, 128.9, 128.8, 128.7, 127.1, 126.7, 126.66, 126.5, 126.46, 125.9, 47.2, 43.4, 43.3, 39.2, 29.0, 28.1; HRMS (TOF) m/z [M + Na]⁺ Calcd for C₁₇H₁₅NNaO₂ 288.0995, found 288.1001; IR (KBr, cm⁻¹) ν 1672, 1639.

N,*N*-Bis(2-methoxyethyl)-2-oxo-2-phenylacetamide (7h). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (126 mg, 95% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.97–7.95 (m, 2H), 7.62–7.58 (m, 2H), 7.50–7.46 (m, 2H), 3.78 (t, *J* = 5.3 Hz, 2H), 3.66 (t, *J* = 5.4 Hz, 2H), 3.54 (t, *J* = 5.3 Hz, 2H), 3.44 (t, *J* = 5.4 Hz, 2H), 3.39 (s, 3H), 3.11 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz) δ 190.7, 167.4, 134.0, 133.2, 129.6, 128.4, 70.2, 70.0, 58.5, 58.2, 47.9, 44.9; HRMS (TOF) m/z [M + Na]⁺ Calcd for C₁₄H₁₉NNaO₄ 288.1206, found 288.1189; IR (KBr, cm⁻¹) ν 1679, 1641.

1-Phenyl-2-(4-(pyridin-2-yl)piperazin-1-yl)ethane-1,2-dione (7i). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (118 mg, 80% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.20–8.19 (m, 1H), 7.99–7.98 (m, 2H), 7.67–7.65 (m, 1H), 7.55–7.50 (m, 3H), 6.71–6.66 (m, 2H), 3.90 (t, *J* = 6.0 Hz, 2H), 3.68 (t, *J* = 6.0 Hz, 2H), 3.57 (t, *J* = 4.0 Hz, 2H), 3.49 (t, *J* = 4.0 Hz, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 191.3, 165.5, 158.7, 147.9, 137.7, 134.9, 133.0, 129.6, 129.0, 114.2, 107.4, 45.47, 45.46, 45.1, 40.9; HRMS (TOF) *m*/*z* [M + H]⁺ Calcd for C₁₇H₁₈N₃O₂ 296.1394, found 296.1393; IR (KBr, cm⁻¹) ν 1678, 1641.

tert-Butyl 4-(2-oxo-2-phenylacetyl)piperazine-1-carboxylate (7j). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (134 mg, 84% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.97–7.95 (m, 2H), 7.69–7.65 (m, 1H), 7.55–7.51 (m, 2H), 3.75 (t, *J* = 4.0 Hz, 2H), 3.56 (t, *J* = 4.0 Hz, 2H), 3.44 (t, *J* = 4.0 Hz, 2H), 3.34 (t, *J* = 4.0 Hz, 2H), 1.47 (s, 9H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.0, 165.4, 154.2, 134.8, 132.8, 129.5, 128.9, 80.4, 45.6, 41.0, 28.2; HRMS (TOF) m/z [M + Na]⁺ Calcd for C₁₇H₂₂N₂NaO₄ 341.1472, found 341.1466; IR (KBr, cm⁻¹) ν 1677, 1650.

1-(4-Hydroxypiperidin-1-yl)-2-phenylethane-1,2-dione (7k). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (84 mg, 72% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.93–7.92 (m, 2H), 7.67–7.64 (m, 1H), 7.54–7.50 (m, 2H), 4.07–3.95 (m, 2H), 3.50–3.44 (m, 2H), 3.18–3.12 (m, 1H), 2.69 (brs, 1H), 1.80–1.49 (m, 4H); ¹³C NMR (CDCl₃,100 MHz) δ 191.7, 165.4, 134.8, 132.8, 129.4, 129.0, 66.1, 43.0, 38.2; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₃H₁₆NO₃ 234.1125, found 234.1120; IR (KBr, cm⁻¹) ν 1677, 1631.

Methyl 1-(2-oxo-2-phenylacetyl)piperidine-4-carboxylate (71). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (121 mg, 88% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.95–7.93 (m, 2H), 7.67–7.63 (m, 1H), 7.54–7.50 (m, 2H), 4.46–4.43 (m, 1H), 3.69 (s, 3H), 3.58–3.55 (m, 1H), 3.13–3.06 (m, 2H), 2.66–2.61 (m, 1H), 2.06–2.04 (m, 1H), 1.91–1.68 (m, 3H); ¹³C NMR (CDCl₃,100 MHz) δ 191.4, 173.9, 165.2, 134.7, 132.8, 129.3, 128.9, 51.8, 45.0, 40.3, 40.2, 28.0, 27.4; HRMS (TOF) m/z [M + H]⁺ Calcd for C₁₅H₁₈NO₄ 276.1230, found 276.1230; IR (KBr, cm⁻¹) ν 1733, 1679, 1642.

N-Isopropyl-2-oxo-2-phenylacetamide (8a). Methyl ketones (0.5 mmol), amines (2.0 mmol), I₂ (0.25 mmol, 50 mol %), TBHP (6.0 equiv, 70% aqueous solution 401 uL), and 2.0 mL of ¹PrOH, were added to a tube under air. The reaction mixture was stirred at 0 °C for 13 h. The reaction mixture was quenched with saturated Na₂S₂O₃ solution, extracted repeatedly with ethyl acetate, and dried over Na₂SO₄. Then, removal of the organic solvent in a vacuum followed by flash silica gel column chromatographic purification afforded product (70 mg, 73% yield) as yellow solid with EtOAc/hexane (1:5) mixtures: ¹H NMR (CDCl₃, 400 MHz) δ 8.32–8.30 (m, 2H), 7.62–7.58 (m, 1H), 7.47–7.43 (m, 2H), 7.05 (s, 1H), 4.22–4.10 (m, 1H), 1.25 (d, *J* = 6.9 Hz, 3H). ¹³C NMR (CDCl₃,75 MHz) δ 188.0, 160.9, 134.2, 133.3, 131.0, 128.3, 41.6, 22.3; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₁H₁₄NO₂ 192, found 192; IR (KBr, cm⁻¹) ν 1671,1643.

N-Benzyl-2-oxo-2-phenylacetamide (8b). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (78 mg, 65% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.33–8.31 (m, 2H), 7.62–7.58 (m, 1H), 7.52 (s, 1H), 7.47–7.43 (m, 2H), 7.35–7.28 (m, 5H), 4.54 (d, *J* = 6.1 Hz, 2H). ¹³C NMR (CDCl₃,75 MHz) δ 187.5, 161.6, 137.0, 134.4, 133.2, 131.1, 128.7, 128.4, 127.8, 127.7, 43.4; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₅H₁₄NO₂ 240, found 240; IR (KBr, cm⁻¹) ν 1683.1648.

N-(2-Cyclohexenylethyl)-2-oxo-2-phenylacetamide (8c). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (70 mg, 54% yield): ¹H NMR (CDCl₃, 400 MHz) δ 8.32–8.30 (m, 2H), 7.63–7.59 (m, 1H), 7.48–7.45 (m, 2H), 7.13 (s, 1H), 5.51 (t, *J* = 3.7 Hz, 1H), 3.49–3.44 (m, 2H), 2.22 (t, *J* = 6.8 Hz, 2H), 2.01–1.96 (m, 4H), 1.64–1.53 (m, 4H). ¹³C NMR (CDCl₃,75 MHz) δ 187.8, 161.7, 134.2, 133.9, 133.2, 131.0, 128.3, 123.9, 37.2, 37.1, 27.8, 25.1, 22.7, 22.2; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₆H₂₀NO₂ 258, found 258; IR (KBr, cm⁻¹) ν 1665.

N-Cyclohexyl-2-oxo-2-phenylacetamide (8d). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (46 mg, 40% yield): ¹H NMR (CDCl₃, 300 MHz) δ 8.32–8.29 (m, 2H), 7.59–7.57 (m, 1H), 7.47–7.43 (m, 2H), 7.08 (s, 1H), 3.87–3.84 (m, 1H), 1.99–1.95 (m, 2H), 1.76–1.62 (m, 3H), 1.38–1.21 (m, 5H); ¹³C NMR (CDCl₃, 75 MHz) δ 188.1, 160. 9, 134.1, 133.3, 131.0, 128.3, 48.3, 32.5, 25.2, 24.6; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₄H₁₈NO₂ 232, found 232; IR (KBr, cm⁻¹) ν 1666.

N-Butyl-2-oxo-2-p-tolylacetamide (8e). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (70 mg, 64% yield): ¹H NMR (CDCl₃, 300 MHz) δ 8.24 (d, *J* = 8.0 Hz, 2H), 7.26 (d, *J* = 8.0 Hz, 2H), 7.20 (s, 1H), 3.41–3.37 (m, 2H), 2.41 (s, 3H), 1.60–1.56 (m, 2H), 1.43–1.36 (m, 2H), 0.94 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 187.4, 162.0, 145.4, 131.2, 130.8, 129.1, 39.0, 31.2, 21.7, 20.0, 13.6; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₃H₁₈NO₂ 220, found 220; IR (KBr, cm⁻¹) ν 1679.

N-Butyl-2-(3-methoxyphenyl)-2-oxoacetamide (8f). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (79 mg, 67% yield): ¹H NMR (CDCl₃, 400 MHz) δ 7.95–7.93 (m, 1H), 7.81 (s, 1H), 7.38–7.34 (m, 1H), 7.24 (s, 1H), 7.17–7.14 (m, 1H), 3.84 (s, 3H), 3.40–3.35 (m, 2H), 1.62–1.54 (m, 2H), 1.44–1.35 (m, 2H), 0.94 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 187.6, 161.9, 159.3, 134.4, 129.3, 123.9, 121.1, 114.5, 55.2, 39.0, 31.2 19.9, 13.5; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₃H₁₈NO₃ 236, found 236; IR (KBr, cm⁻¹) ν 1662.

N-Butyl-2-(4-fluorophenyl)-2-oxoacetamide (8g). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (72 mg, 64% yield): ¹H NMR (CDCl₃, 300 MHz) δ 8.44 (d, *J* = 6.2 Hz, 2H), 7.22 (s, 1H), 7.15 (d, *J* = 6.2 Hz, 2H), 3.40–3.38 (m, 2H), 1.59–1.57 (m, 2H), 1.41–1.39 (m, 2H), 0.95 (t, *J* = 7.9 Hz, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 186.0, 168.2, 164.8, 161.5, 134.2, 134.1, 129.8, 115.8, 115.5, 39.1, 31.2, 20.0, 13.6; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₂H₁₅FNO₂ 224, found 224; IR (KBr, cm⁻¹) ν 1656.

N-Butyl-2-oxo-2-(thiophen-2-yl)acetamide (8h). Purified by column chromatography (EtOAc:hexane 1:5). Yellow solid (67 mg, 63% yield): ¹H NMR (CDCl₃, 300 MHz) δ 8.41 (d, *J* = 2.7 Hz, 1H), 7.83 (d, *J* = 2.7 Hz, 1H), 7.36 (s, 1H), 7.20–7.18 (m, 1H), 3.42–3.36 (m, 2H), 1.60–1.55 (m, 2H), 1.44–1.36 (m, 2H), 0.95 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 178.4, 160.6, 138.5, 137.8, 136.6, 128.0, 39.2, 31.1, 19.9, 13.6; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₁₂H₁₄NO₂S 212, found 212; IR (KBr, cm⁻¹) ν 1685, 1648.

1-(4-Bromophenyl)-2-morpholinoethanone (4). A mixture of 2-bromo-1-(4-bromophenyl)ethanone (1 mmol), morpholine (2 equiv) in dichloromethane was heated at 25 °C for 6 h. Removal of the organic solvent in a vacuum followed by flash silica gel column chromatographic purification afforded product (260 mg, 92% yield) as yellow solid with EtOAc/hexane (1:5) mixtures: ¹H NMR (CDCl₃, 400 MHz) δ 7.89 (d, *J* = 8.0 Hz, 2H), 7.61 (d, *J* = 8.0 Hz, 2H), 3.77 (t, *J* = 4.0 Hz, 4H), 3.76 (s, 2H), 2.60 (t, *J* = 4.0 Hz, 4H); ¹³C NMR (CDCl₃, 100 MHz) δ 195.2, 134.5, 131.8, 129.7, 128.5, 66.7, 64.8, 53.8; HRMS (TOF) *m/z* [M + H]⁺ Calcd for C₁₂H₁₅⁷⁹BrNO₂ 284.0281, found 284.0281; C₁₂H₁₅⁸¹BrNO₂ 284.0266, found 286.0262; IR (KBr, cm⁻¹) ν 1689.

1-(4-Bromophenyl)-2-iodoethanone (5). A mixture of 1-(4bromophenyl)ethanone (1 mmol), CuO (1 equiv), and iodine (1 equiv) in methanol was heated at 65 °C for 1 h. Removal of the organic solvent in a vacuum followed by flash silica gel column chromatographic purification afforded product (292 mg, 90% yield) as colorless solid with EtOAc/hexane (1:20) mixtures: ¹H NMR (CDCl₃, 400 MHz) δ 7.85 (d, *J* = 8.4 Hz, 2H), 7.63 (d, *J* = 8.4 Hz, 2H), 4.32 (s, 2H); ¹³C NMR (CDCl₃,100 MHz) δ 191.8, 132.13, 132.08, 130.5, 129.1, 1.2; ESI-MS *m*/*z* [M + H]⁺ Calcd for C₈H₇⁻⁹BrIO 325, found 325; C₈H₇⁸¹BrIO 327, found 327; IR (KBr, cm⁻¹) ν 1688.

ASSOCIATED CONTENT

S Supporting Information

Copies of ¹H NMR and ¹³C NMR spectra for all products. This material is available free of charge via the Internet at http:// pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: wanxb@suda.edu.cn.

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REFERENCES

(1) (a) Serullas, G. Ann. Chim. **1822**, 20, 165. (b) Fuson, R. C.; Bull, B. A. Chem. Rev. **1934**, 15, 275.

(2) For recent reviews on this topic, see: (a) Li, C.-J. Acc. Chem. Res. 2009, 42, 335. (b) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Chem. Commun. 2010, 46, 677. (c) Dobereiner, G. E.; Crabtree, R. H. Chem. Rev. 2010, 110, 681. (d) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624. (e) Copéret, C. Chem. Rev. 2010, 110, 656. (f) Lyons, T. W.; Sanford, M. S. Chem. Rev. 2010, 110, 1147. (g) Mkhalid, I. A. I.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. Chem. Rev. 2010, 110, 890. (h) Doyle, M. P.; Duffy, R.; Ratnikov, M.; Zhou, L. Chem. Rev. 2010, 110, 704. (i) Gunay, A.; Theopold, K. H. Chem. Rev. 2010, 110, 1060. (j) Balcells, D.; Colt, E.; Eisenstein, O. Chem. Rev. 2010, 110, 749. (k) Bellina, F.; Rossi, R. Chem. Rev. 2010, 110, 1082. (l) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Chem. Rev. 2011, 111, 1293. (m) Yeung, C. S.; Dong, V. M. Chem. Rev. 2011, 111, 1215. (n) Le Bras, J.; Muzart, J. Chem. Rev. 2011, 111, 1170. (o) Ackermann, L. Chem. Rev. 2011, 111, 1315.

(3) For selected examples, see: (a) Ocain, T. D.; Rich, D. H. J. Med. Chem. **1992**, 35, 451. (b) Knust, H.; Nettekoven, M.; Pinard, E.; Roche, O.; Rogers-Evans, M. PCT Int. Appl. WO 2009016087, 2009. (c) Sheha, M. M.; Mahfouz, N. M.; Hassan, H. Y.; Youssef, A. F.; Mimoto, T.; Kiso, Y. Eur. J. Med. Chem. **2000**, 35, 887.

(4) For selected examples, see: (a) Jesuraj, J. L.; Sivaguru, J. Chem. Commun. 2010, 46, 4791. (b) Zhang, Z.; Zhang, Q.; Ni, Z.; Liu, Q. Chem. Commun. 2010, 46, 1269. (c) Sai, K. K. S.; Esteves, P. M.; da Penha, E. T.; Klumpp, D. A. J. Org. Chem. 2008, 73, 6506. (d) Tomita, D.; Yamatsugu, K.; Kanai, M.; Shibasaki, M. J. Am. Chem. Soc. 2009, 131, 6946. (e) Yang, L.; Wang, D.-X.; Huang, Z.-T.; Wang, M.-X. J. Am. Chem. Soc. 2009, 131, 10390. (f) Xu, H.; Wolf, C. Angew. Chem., Int. Ed. 2011, 50, 12249. (g) Wang, R.; Chen, C.; Duesler, E.; Mariano, P. S; Yoon, U. C. J. Org. Chem. 2004, 69, 1215.

(5) (a) Chen, J.; Cunico, R. F. J. Org. Chem. 2004, 69, 5509.
(b) Murata, S.; Suzuki, K.; Miura, M.; Nomura, M. J. Chem. Soc., Perkin Trans. 1 1990, 361. (c) Singh, R. P.; Shreeve, J. M. J. Org. Chem. 2003, 68, 6063.

The Journal of Organic Chemistry

(6) (a) Kobayashi, T.; Tanaka, M. J. Organomet. Chem. 1982, 233,
C64. (b) Ozawa, F.; Soyama, H.; Yanagihara, H.; Aoyama, I.; Takino,
H.; Izawa, K.; Yamamoto, T.; Yamamoto, A. J. Am. Chem. Soc. 1985,
107, 3235. (c) Mizushima, E.; Hayashi, T.; Tanaka, M. Green Chem.
2001, 3, 76. (d) Tsukada, N.; Ohba, Y.; Inoue, Y. J. Organomet. Chem.
2003, 687, 436. (e) Iizuka, M.; Kondo, Y. Chem. Commun. 2006, 1739.
(f) Rahman, Md. T.; Fukuyama, T.; Kamata, N.; Sato, M.; Ryu, I.
Chem. Commun. 2006, 2236. (g) Murphy, E. R; Martinelli, J. R.;
Zaborenko, N.; Buchwald, S. L.; Jensen, K. F. Angew. Chem., Int. Ed.
2007, 46, 1734. (h) Liu, J.; Zhang, R.; Wang, S.; Sun, W.; Xia, C. Org.
Lett. 2009, 11, 1321. (i) de la Fuente, V.; Godard, C.; Zangrando, E.;
Claver, C.; Castillón, S. Chem. Commun. 2012, 48, 1695.

(7) (a) Zhang, C.; Jiao, N. J. Am. Chem. Soc. 2010, 132, 28.
(b) Zhang, C.; Xu, Z.; Zhang, L.; Jiao, N. Angew. Chem., Int. Ed. 2011, 50, 11088. (c) Du, F.-T.; Ji, J.-X. Chem. Sci. 2012, 3, 460. (d) Zhang, C.; Zong, X.; Zhang, L.; Jiao, N. Org. Lett. 2012, 14, 3280.

(8) (a) Chen, J. J.; Deshpande, S. V. *Tetrahedron Lett.* 2003, 44, 8873.
(b) El Kaïm, L.; Pinot-Périgord, E. *Tetrahedron* 1998, 54, 3799.
(c) Grassot, J.-M.; Masson, G.; Zhu, J. *Angew. Chem., Int. Ed.* 2008, 47, 947.

(9) (a) See ref 3a. (b) Semple, J. E.; Owens, T. D.; Nguyen, K.; Levy,
O. E. Org. Lett. 2000, 2, 2769. (c) Banfi, L.; Guanti, G.; Riva, R. Chem.
Commun. 2000, 985. (d) Chiou, A.; Markidis, T.; Constantinou-Kokotou, V.; Verger, R.; Kokotos, G. Org. Lett. 2000, 2, 347.
(e) Nakamura, M.; Inoue, J.; Yamada, T. Bioorg. Med. Chem. Lett.
2000, 10, 2807. (f) Xu, P.; Lin, W.; Zhou, X. Synthesis 2002, 1017.

(10) (a) Buckley, T. F.; Rapoport, H. J. Am. Chem. Soc. **1982**, 104, 4446. (b) Papanikos, A.; Rademann, J.; Medal, M. J. Am. Chem. Soc. **2001**, 123, 2176. (c) Chen, J.; Chen, X.; Bois-Choussy, M.; Zhu, J. J. Am. Chem. Soc. **2006**, 128, 87.

(11) (a) Takahashi, K.; Shibasaki, K.; Ogura, K.; Iida, H. *Chem. Lett.* **1983**, 859. (b) Yang, Z.; Zhang, Z.; Meanwell, N. A.; Kadow, J. F.; Wang, T. *Org. Lett.* **2002**, *4*, 1103. (c) Zhu, J.; Wong, H.; Zhang, Z.; Yin, Z.; Kadow, J. F.; Meanwell, N. A.; Wang, T. *Tetrahedron Lett.* **2005**, *46*, 3587.

(12) (a) Wasserman, H. H.; Ho, W. B. J. Org. Chem. 1994, 59, 4364.
(b) Wasserman, H. H.; Petersen, A. K.; Xia, M. Tetrahedron 2003, 59, 6771.

(13) (a) Padwa, A.; Koehler, K. F. J. Chem. Soc., Chem. Commun. 1986, 789. (b) Konstantinova, L. S.; Bol'shakov, O. I.; Obruchnikova, N. V.; Golova, S. P.; Nelyubina, Y. V.; Lyssenko, K. A.; Rakitin, O. A. Tetrahedron 2010, 66, 4330.

(14) (a) Rehwald, M.; Gewald, K.; Lankau, H.-J.; Unverferth, K. *Heterocycles* **1997**, *45*, 483. (b) Ji, M.; Chen, J.; Ding, K.; Wu, X.; Varady, J.; Levantb, B.; Wang, S. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 1701.

(15) (a) Hong, R.; Feng, J.; Hoen, R.; Lin, G.-q. Tetrahedron 2001, 57, 8685. (b) Chen, H.-R.; Guo, X.-K.; Zhong, X.-B. Chin. J. Chem. 2006, 24, 1411.

(16) Recently, we and others reported metal free oxidation reaction using TBHP, see: (a) Chen, L.; Shi, E.; Liu, Z.; Chen, S.; Wei, W.; Li, H.; Xu, K.; Wan, X. Chem.-Eur. J. 2011, 17, 4085. (b) Uyanik, M.; Suzuki, D.; Yasui, T.; Ishihara, K. Angew. Chem., Int. Ed. 2011, 50, 5331. (c) Froehr, T.; Sindlinger, C. P.; Kloeckner, U.; Finkbeiner, P.; Nachtsheim, B. J. Org. Lett. 2011, 13, 3754. (d) Wei, W.; Zhang, C.; Xu, Y.; Wan, X. Chem. Commun. 2011, 47, 10827. (e) Ma, L.; Wang, X.; Yu, W.; Han, B. Chem. Commun. 2011, 47, 11333. (f) Tang, R.-Y.; Xie, Y.-X.; Xie, Y.-L.; Xiang, J.-N.; Li, J.-H. Chem. Commun. 2011, 47, 12867. (g) Xie, J.; Jiang, H.; Cheng, Y.; Zhu, C. Chem. Commun. 2012, 48, 979. (h) Chen, S.; Xu, Y.; Wan, X. Org. Lett. 2011, 13, 6152. (i) Kloeckner, U.; Weckenmann, N. M.; Nachtsheim, B. J. Synlett 2012, 97. (j) Gao, M.; Yang, Y.; Wu, Y.-D.; Deng, C.; Shu, W.-M.; Zhang, D.-X.; Cao, L.-P.; She, N.-F.; Wu, A.-X. Org. Lett. 2010, 12, 4026. (k) Xue, W.-J.; Li, Q.; Zhu, Y.-P.; Wang, J.-G.; Wu, A.-X. Chem. Commun. 2012, 48, 3485. (1) Zhu, Y.-p.; Liu, M.-c.; Jia, F.-c.; Yuan, J.j.; Gao, Q.-h.; Lian, M.; Wu, A.-x. Org. Lett. 2012, 14, 3392. (m) He, T.; Yu, L.; Zhang, L.; Wang, L.; Wang, M. Org. Lett. 2011, 13, 5016. (n) Shi, E.; Shao, Y.; Chen, S.; Hu, H.; Liu, Z.; Zhang, J.; Wan, X. Org.

Lett. 2012, 14, 3384. (o) Yan, Y.; Zhang, Y.; Feng, C.; Zha, Z.; Wang, Z. Angew. Chem., Int. Ed. 2012, 51, 8077.

(17) For iminium intermediate generated in situ from amine in the presence of TBHP, see: (a) Li, Z.; Li, C.-J. J. Am. Chem. Soc. **2004**, 126, 11810. (b) Li, Z.; Li, C.-J. Org. Lett. **2004**, 6, 4997. (c) Li, Z.; Li, C.-J. J. Am. Chem. Soc. **2005**, 127, 3672. (d) Li, Z.; Li, C.-J. Eur. J. Org. Chem. **2005**, 3173.

(18) For nucleophilic attack of iminium intermediate by H_2O , see: (a) Dotzauer, M.; Eisfeld, W.; Vilsmaier, E.; Fröhlich, K.; Bergsträsser, U.; Tetzlaff, C. J. Org. Chem. **1996**, 61, 8526. (b) Yin, W.; Mitra, K.; Stearns, R. A.; Baillie, T. A.; Kumar, S. Biochemistry **2004**, 43, 5455. (c) Wang, J.; Xu, H.; Gao, H.; Su, C.-Y.; Zhao, C.; Phillips, D. L. Organometallics **2010**, 29, 42. (d) Baslé, O.; Li, C.-J. Org. Lett. **2008**, 10, 3661.

(19) For oxidation of α -hydroxy amine, see: (a) Tillack, A.; Rudloff, I.; Beller, M. Eur. J. Org. Chem. 2001, 523. (b) Yoo, W.-J.; Li, C.-J. J. Am. Chem. Soc. 2006, 128, 13064. (c) Ekoue-Kovi, K.; Wolf, C. Org. Lett. 2007, 9, 3429. (d) Reddy, K. R.; Maheswari, C. U.; Venkateshwar, M.; Kantam, M. L. Eur. J. Org. Chem. 2008, 3619.

(20) For hydroxide anion generated in situ from TBHP, see: (a) Kharasch, M. S.; Pauson, P.; Nudenberg, W. J. Org. Chem. 1953, 18, 322. (b) Barton, D. H. R.; Le Gloahec, V. N.; Patin, H.; Launay, F. New J. Chem. 1998, 559. (c) Jones, C. M.; Burkitt, M. J. J. Am. Chem. Soc. 2003, 125, 6946. (d) McLaughlin, E. C.; Choi, H.; Wang, K.; Chiou, G.; Doyle, M. P. J. Org. Chem. 2009, 74, 730.